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Ksp Problems – Chemistry Name: _____ 1) The value of Ksp of AgCl is 1.8×10^{-10} . What would be the molar concentration of Ag⁺ and Cl⁻ in pure water placed in contact with solid AgCl(s)? $K_{sp} = [Ag^+][Cl^-]$
 $1.8 \times 10^{-10} = [x][x]$ $x = 0.000013416$ M or 1.3416×10^{-5} M 2) The molar solubility of PbCl ...

Ksp Problems – Chemistry

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3—3 Practice Problems 1. How many protons and electrons are 8. What is the name of element that has present in a vanadium atom? atoms that contain 25 protons? 2. How many protons and electrons are 9. What is the name of the element that has present in a nitrogen atom? atoms that contain 82 protons? 3. How many protons and electrons are 10.

Name Date Class 3—3 Practice Problems - Weebly

0.710 kg of sodium carbonate (washing soda), Na₂CO₃, in 10.0 kg of water—a saturated solution at 0 ° C;
125 g of NH₄NO₃ in 275 g of water—a mixture used to make an instant ice pack; 25 g of Cl₂ in 125 g of dichloromethane, CH₂Cl₂; 0.372 g of histamine, C₅H₉N, in 125 g of chloroform, CHCl₃

8.3: Concentrations of Solutions (Problems) - Chemistry ...

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