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numbers of neutrons are called _____. answer choices
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Chapter 18: Properties of Atoms and the Periodic

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Table ...

Chapter 1. Atoms Student Objectives 1.1 A Particulate View of the World: Structure Determines Properties □ Define atoms, molecules, and the science of chemistry. □ Represent a simple molecule, water, using spheres as atoms. 1.2 Classifying Matter: A Particulate View

Chapter 1. Atoms - Solutions Manual - Test Bank
Electrons removed from atoms always increase the stability of the atom. The energy of electrons decreases with increasing distance from the nucleus. Electrons are considered valence electrons if...

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Atoms, Elements & the Periodic Table - Practice Test

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Chapter 17 Properties of Atoms and the Periodic Table

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B) Atoms of different elements have different properties. C) The volumes of gases that combine are in small whole number ratios. D) Atoms combine in fixed ratios of whole numbers when they form compounds. E) In chemical reactions, atoms are not created or destroyed. Ans: C Introduction to Chemistry 3rd Edition Bauer Test Bank

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Chapter 2: Atoms, Ions, and the Periodic Table - My Test Bank

A)chlorine is a gas and sodium is a solid B)chlorine is more metallic than sodium C)chlorine has a greater ionization energy than sodium does D)chlorine has a greater electron affinity than sodium does E)chlorine is bigger than sodium 46) Of the elements below, _____ is the most metallic.

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Concepts: Terms in this set (17) atomic mass unit. approximately 1.67×10^{-24} g. mass number. the sum of the number of neutrons and protons in an atoms nucleus.

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Atoms Elements And The Periodic Table Chapter Test Answer ...

Atoms are the most fundamental of all particles that can have an independent existence. Sometimes two or more atoms combine to form a big particle. When atoms combine, they form molecules. When the particles of a substance contain only one type of atoms, that substance is called an element. In elements the smallest particle that exist may be atoms or molecules.

Chapter ATOMS AND MOLECULES

Atoms Elements And The Periodic Table Chapter

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Electrons removed from atoms always increase the stability of the atom. The energy of electrons decreases with increasing distance from the nucleus. Electrons circle the nucleus in predictable orbits like the Earth around the sun. Chapter 10 Atomic Structure And The Periodic Table Answer Key ...

Chapter Test Atoms And The Periodic Table Answers
b. properties. d. symbols. ____ 7. Magnesium (Mg) is located to the right of sodium (Na) because Mg has a.

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fewer protons. c. no protons. b. no neutrons. d. more protons. ____8. As you move from left to right across the periodic table, elements a. become less metallic. c. have a lower atomic weight. b. have a lower atomic number. d. become more metallic. ____9.

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