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sphere of homogeneous solution. 6.2

EXPLAINING SOLUTIONS PRACTICE (Page 273)

Understanding Concepts 1. Most molecules are polar. The four categories of polar molecules are: AB (e.g., $HCl(g)$, $CO(g)$) $N \times A_y$ (e.g., $NH_3(g)$, $NF_3(g)$) $O \times A_y$ (e.g., $H_2O(l)$, $OCl_2(g)$) $C \times A_y B_z$ (e.g., $CHCl_3(l)$, $C_2H_5OH(l)$) 2.

6.1 DEFINING A SOLUTION

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