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An ideal gas is a hypothetical gas consisting of identical particles with zero volume and with no intermolecular forces. Addition-ally, the constituent atoms or molecules undergo perfectly elastic collisions with the walls of the container. An ideal gas can con-sist of molecules (e.g. carbon dioxide molecules, CO₂) or atoms (e.g. neon atoms, Ne). Real gases do not exhibit these exact prop-

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Plants convert carbon dioxide, water, and energy from the sun into sugar (C₆H₁₂O₆) and dioxygen in the process of photosynthesis.

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